

## METAPHOSPHATE AND AEROBIC CO<sub>2</sub> IN CHLORELLA

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### 1. Introduction

A striking parallelism between the metaphosphate and the "aerobic CO<sub>2</sub>" of Chlorella has been observed.

If the photolyte [1] of photosynthesis is built up in Chlorella, the first intermediate is the "aerobic CO<sub>2</sub>". It is called "aerobic", because the fixation of this CO<sub>2</sub> requires not only CO<sub>2</sub> but also O<sub>2</sub>. The equilibrium of the fixation of the aerobic CO<sub>2</sub> is reached at 20° in 30 minutes. At the same rate the aerobic CO<sub>2</sub> dissociates if the CO<sub>2</sub> or the O<sub>2</sub> is removed from the gas mixture in the presence of which the aerobic CO<sub>2</sub> was formed.

We have discovered that the highly polymeric metaphosphate of Chlorella increases, when the "aerobic CO<sub>2</sub>" is formed and that the metaphosphate decreases, when the "aerobic CO<sub>2</sub>" dissociates on the removal of the CO<sub>2</sub> or the O<sub>2</sub>.

### 2. Isolation of the metaphosphate

Chlorella suspensions are acidified at room temperature with acetic acid to a final concentration of 0.2 N. The supernatant is removed on the centrifuge and the sediment is extracted with methanol until the residue is white. The white residue is extracted 3 times with 0.1 N aqueous NH<sub>3</sub>. The extract contains the metaphosphate.

The extract is acidified with acetic acid to pH 4. A considerable precipitate is removed on the centrifuge. To the supernatant HCl is added to pH 2. A small precipitate is removed on the centrifuge. To the supernatant a great excess of BaCl<sub>2</sub> is added, which precipitates the metaphosphate as Ba-salt. This is redissolved in NaOH and re-precipitated at pH 2 with a

great excess of BaCl<sub>2</sub>. After drying in high vacuum at 60° it contained 41.1% Ba and 18.7% P, as expected for Ba/2 PO<sub>3</sub> + H<sub>2</sub>O.

The Ba-salt was free of organic substances and of nitrogen. Chromatography according to Rossel [2] showed, that it contained no Na<sub>5</sub>P<sub>2</sub>O<sub>10</sub>, no pyrophosphate, only traces of orthophosphate and only non-migrating phosphate. All the phosphate of the Ba-salt was split to orthophosphate by heating 30 minutes in 2 N HCl. After the removal of the Ba with H<sub>2</sub>SO<sub>4</sub> the supernatant gave the toluidine-blue test according to Wiame [3] and precipitated dialysed serum proteins in diluted acetic acid at room temperature according to Mann [4]. Thus the metaphosphate extracted by 0.1 N NH<sub>3</sub> from Chlorella was highly polymeric.

### 3. Yield of metaphosphate

From 15000 µl Chlorella 38.5 mg dried Ba-salt were obtained, containing 232 micromoles P when the cells had been pretreated at 20° with 20 Vol % CO<sub>2</sub> in air. This means a yield of 1.55 micromoles metaphosphate P from 100 µl cells saturated with the CO<sub>2</sub>-air mixture. When, on the other hand, the cells were pretreated with CO<sub>2</sub> but without O<sub>2</sub> and therefore were free of "aerobic CO<sub>2</sub>", the yield of metaphosphate was only 1.1 micromoles P from 100 µl Chlorella. No pre-treatment of cells could raise the yield of the metaphosphate above 1.1 micromoles of P from 100 µl cells except the saturation with CO<sub>2</sub> + O<sub>2</sub>, that is except the synthesis of the "aerobic CO<sub>2</sub>".

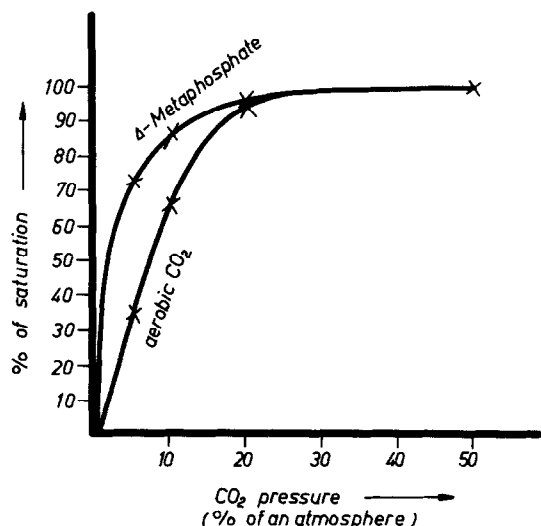


Fig. 1. 400  $\mu$ l of *Chlorella*, suspended in 6 ml of a neutral solution of 0.02 M  $\text{MgSO}_4$  and 0.034 M NaCl were shaken at 20° in the dark with different mixtures of  $\text{CO}_2$  and air. When, after 30 minutes, the equilibrium concentrations of the "aerobic  $\text{CO}_2$ " had been reached, metaphosphate was determined as described in the text,  $\Delta$  metaphosphate being  $\times$  metaphosphate in  $\text{CO}_2$  and air minus metaphosphate in air. In a parallel experiment the "aerobic  $\text{CO}_2$ " was determined with the fluoride method [1].

#### 4. Parallelism between metaphosphate and aerobic $\text{CO}_2$

When the increments of the metaphosphate and the aerobic  $\text{CO}_2$  were determined at different partial

pressures of  $\text{CO}_2$  in air, very similar curves were obtained, as shown in fig. 1. Thus the absolute increments of the metaphosphate and the aerobic  $\text{CO}_2$  were equal at high  $\text{CO}_2$  pressures, that is

$$\Delta \text{ metaphosphate} = \text{aerobic } \text{CO}_2$$

Both changes are completely reversible. If the  $\text{CO}_2$  or the  $\text{O}_2$  are removed, both increments disappear and the rates of disappearances are equal. It is not claimed that the aerobic  $\text{CO}_2$  is a metaphosphate compound, but the findings reveal a close connection between metaphosphate and the "aerobic  $\text{CO}_2$ ".

#### 5. Light and metaphosphate

Light has no effect on the metaphosphate in *Chlorella*, if the  $\text{CO}_2$  and  $\text{O}_2$  pressures are kept constant in the dark and in the light. Only when light removes the  $\text{CO}_2$  by photosynthesis, does the metaphosphate disappear together with the "aerobic  $\text{CO}_2$ ".

#### References

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